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UTILIZATION OF WASTE HEAT FROM WASTE HEAT RECOVERY BOILER THROUGH A WATER TUBE HEAT EXCHANGER Rahul Deore^{*1} & Nitesh Rane²

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ABSTRACT

Rotary kiln the there is lots of amount of heat is generated during the combustion of coal and that heat to be utilized before throwing it into the environment. Waste heat recovery boiler is quite suitable to utilize that heat from kiln but there is region in boiler where the heat is not utilized efficiently. So, the approach should be made for same purpose. But it is necessary to use waste heat in such a form to make system efficient and productive. Then we have decided to design the heat exchanger for that particular region. But condensate extraction pump (CEP) and deaerator works on low temperature and pressure, so it is necessary to design heat exchanger in same way. Water will be feed through condensate extraction pump (CEP) and condensate water heated in heat exchanger and then goes to deaerator. This water will reduce the fuel consumption and will make system cost effective.

KEYWORDS: Waste heat recovery boiler (WHRB), heat exchanger, Condensate extraction pump (CEP), deaerator.

1. INTRODUCTION

Rotary kiln process

Rotary kiln is nothing but cylindrical long vessel placed horizontally with some inclination to guide the material flow in downward side. In which material feed from its inlet side and transferred to outlet side by rotating it. In material the mixture of iron ore high fixed carbon coal and dolomite is feed. But for reaction it is necessary to inject coal fines from outlet side to gain temperature and reaches to reaction temperature. By this way lots heat is generated during the process and that heat is generated during the process and that can be utilized in waste heat recovery boiler.

Water tube heat exchanger process

Waste heat recovery boiler is a water tube boiler and will be used as heat recovery boiler but with some limitation it cannot be utilized 100%. As per study we found that some of the waste heat can be utilized in ESP inlet and economizer outlet. This waste heat can be utilized by a designing water tube heat exchanger. Water tube heat exchanger will be installed in between electrostatic precipitator (ESP) inlet and economizer outlet and condensate water will feed from Condensate extraction pump (CEP) and water heated from 60 °C up to 125 °C. The temperature difference is 65 °C. Water will go to deaerator. It is basically sensible type of heat exchanger which gains sensible heat. By this is way we have tried to make system efficient and effective.

Waste heat recovery systems

- Recuperator.
- Economizer.
- Regenerator.
- Condenser.
- Heat pipe.
- Waste heat recovery boiler.

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2. DESIGN

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1. Calculate number of heat exchanger tubes

Total mass flow rate of CEP condensate water $\ensuremath{m_w}$	175957 kg/hr	
Margin mass flow rate of CEP condensate water $\ensuremath{m_w}$	12996 kg/hr	
Margin mass flow rate of CEP condensate water $\ensuremath{m_w}$	3.61 kg/sec	
Water temperature at heat exchanger inlet T_{c1}	333 k	
Water temperature at heat exchanger inlet T_{c1}	60 °C	
Mass flow rate of flue gas from 500TPD kiln m_{fg}	161640 kg/hr	
Mass flow rate of flue gas from 500TPD kiln m_{fg}	44.90 kg/sec	
Flue gas temperature at economizer outlet T _{h1}	473 k	
Flue gas temperature at economizer outlet T _{h1}	200 °C	
Flue gas temperature at ESP inlet T _{h2}	453 k	
Flue gas temperature at ESP inlet Th2	180 °C	

Table 1. Given data for water tube heat exchanger

2. Heat transfer rate of flue gas (Q)

$$\begin{split} & Q = m_{fg} \times c_{p \ (fg)} \times (T_{h_1} - T_{h_2}) \\ & Q = 44.90 \times 1.097 \times (473 - 453) \\ & Q = 985.106 \ kj/sec \end{split}$$

3. Heat loss by hot flue gases and heat gained by condensate water $m_{1} \times c_{1} \times A^{T} = m_{2} \times c_{2} \times A^{T}$

$$\begin{split} m_{fg} &\times c_{p\,(fg)} \times \Delta T = m_w \times c_{p\,(w)} \times \Delta T \\ m_{fg} &\times c_{p\,(fg)} \times (T_{h_1} - T_{h_2}) = m_w \times c_{p\,(w)} \times (T_{c_2} - T_{c_1}) \\ 44.90 \times 1.097 \times (473 - 453) = 3.61 \times 4.185 \times (T_{c_2} - 333) \\ T_{c_2} &= 398 \text{ K} (125 \text{ °C}) \end{split}$$

4. Log mean temperature difference (ΔT)

ference (Δ1) $\Delta T = \frac{(T_{h_1} - T_{c_2}) - (T_{h_2} - T_{c_1})}{\ln \left[\frac{T_{h_1} - T_{c_2}}{T_{h_2} - T_{c_1}}\right]}$ $\Delta T = \frac{(473 - 398) - (453 - 333)}{\ln \left[\frac{473 - 398}{453 - 333}\right]}$ $\Delta T = 368.74 \text{ K (95.74 °C)}$ $\Delta T_{lm} = F \times \Delta T_{lm (counter flow)}$ $\Delta T_{lm} = 0.99 \times 95.74$ $\Delta T_{lm} = 367.78 \text{ k (94.78 °C)}$

Table 2. Find out data for water tube heat exchanger

Heat transfer coefficient of water $h_{i(w)}$	$5250.74 \text{ w/m}^2\text{k}$
Heat transfer coefficient of flue gas htttial (fg)	140.31 w/m ² k
Overall heat transfer coefficient Uo	133.84 w/m ² k
Capacity ratio (C)	0.3
NTU	0.74
Surface area of tube (A)	77.65 m ²

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5. Number of tube(N)

 $A = \pi \times d_o \times L \times N \times P$ 77.65 = 3.14 × 0.024 × 1.110 × N × 6 N = 156

6. Effectiveness of counter flow heat exchanger (ϵ)

$$\epsilon = \frac{1 - e^{[-NTU(1-c)]}}{1 - ce^{[-NTU(1-c)]}}$$

$$\epsilon = \frac{1 - e^{[-0.74(1-0.3)]}}{1 - 0.3 \times e^{[-0.74(1-0.3)]}}$$

$$\epsilon = 0.5$$

Number of tubes (N)	156	
Passes Or Pitch (S _L)	Longitudinal (6)	
Passes Or Pitch (S _T)	Transverse (26)	
Length of tube	1.110 m	
Tube OD	24 mm	
Tube ID	16 mm	
Tube thickness	8 mm	
Arrangement	Inline	
Flow	Cross	
Weight of heat exchanger with water	457 kg	
Water holding capacity	58.12 litre	

Table 2.Specification of water tube heat exchanger



Water tube heat exchanger (Side view)

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Water tube heat exchanger (Front view)

7. Calculate pressure drop (Δp)

7.1. Calculate pressure drop between CEP to HE (Δp_1)

$$\frac{p_1}{\rho \times g} + \frac{v_1^2}{2 \times g} + z_1 = \frac{p_2}{\rho \times g} + \frac{v_2^2}{2 \times g} + z_2 + h_{L3 \text{ (total)}}$$

$$\frac{700000}{983 \times 9.81} + \frac{0.75^2}{2 \times 9.81} + 0 = \frac{p_2}{983 \times 9.81} + \frac{0.75^2}{2 \times 9.81} + 14 + (3.3808)$$

$$p_2 = 5.3239 \text{ bar}$$

$$\Delta p_1 = p_2 - p_1$$

$$\Delta p_1 = 7 - 5.3239$$

$$\Delta p_1 = 1.6761 \text{ bar}$$

7.2. Calculate pressure drop at HE (
$$\Delta p_2$$
)

$$\frac{p_2}{\rho \times g} + \frac{v_2^2}{2 \times g} + z_2 = \frac{p_3}{\rho \times g} + \frac{v_3^2}{2 \times g} + z_3 + h_{L \text{(total)}}$$

$$\frac{532390}{983 \times 9.81} + \frac{0.75^2}{2 \times 9.81} + 0 = \frac{p_3}{983 \times 9.81} + \frac{0.75^2}{2 \times 9.81} + 0 + (9.1546)$$

$$p_3 = 4.4411\text{bar}$$

$$\Delta p_2 = p_2 - p_3$$

$$\Delta p_2 = 5.3239 - 4.4411$$

$$\Delta p_2 = 0.9928 \text{ bar}$$

$$\Delta p_2 = 0.8828 \text{ bar}$$

7.3. Calculate pressure drop between HE to deaerator
$$(\Delta p_3)$$

$$\frac{p_3}{p_3} + \frac{v_3^2}{p_4} + z_4 = \frac{p_4}{p_4} + \frac{v_4^2}{p_4} + z_5 + h$$

$$\frac{1}{p \times g} + \frac{1}{2 \times g} + \frac{1}{2 \times g} + \frac{1}{p \times g} + \frac{1}{2 \times g} + \frac{1}{2 \times g} + \frac{1}{2 \times g} + \frac{1}{1} + \frac{1}{1$$

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$$\begin{array}{l} \Delta p_3 = 4.4315 - 3.3368 \\ \Delta p_3 = 1.0947 \ \text{bar} \end{array}$$

7.4. Calculate total pressure drop (Δp)

$$\Delta p = p_1 - p_4$$

 $\Delta p = 7 - 3.3368$
 $\Delta p = 3.6632$ bar

- 8. Calculate Total pressure drop across tube and across duct (Δp)
- 8.1. Total pressure drop across tube (Δpd_o)

$$\Delta p = N_L \times f \times x \times \left(\frac{\rho \times v_{max}^2}{2}\right)$$
$$\Delta p = 4 \times 0.5 \times 1 \times \left(\frac{0.748 \times 16^2}{2}\right)$$
$$\Delta p = 287.23 \text{ n/m}^2$$
$$\Delta p = 29.2893 \text{ mmwc}$$

8.2 Pressure drop across duct (Δp_{Duct})

$$\Delta p = \frac{f \times L}{m} \left[\frac{V}{4.04} \right]^2$$
$$\Delta p = \frac{0.005 \times 2.36}{0.3516} \left[\frac{8}{4.04} \right]^2$$
$$\Delta p = 0.1315 \text{ mmwc}$$

- 8.3 Total pressure drop across tube and across duct (Δp) $\Delta p = \text{pressure drop across tube outside + pressure drop in rectangle duct}$ $\Delta p = 29.2893 + 0.1315$ $\Delta p = 29.4208 \text{ mmwc}$
- 9. Thermal Analysis



Model (C4) Mesh 1 of 2

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Model (C4) Mesh 2 of 2



Model (C4) steady state thermal (C5) temperature 1 of 2



Model (C4) steady state thermal (C5) temperature 2 of 2

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10. CFD Analysis



Model (C4) steady state thermal (C5) solution (C6) temperature profile 1 of 2



Model (C4) steady state thermal (C5) solution (C6) temperature profile 2 of 2

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Model (C4) steady state thermal (C5) solution (C6) pressure profile



Model (C4) steady state thermal (C5) solution (C6) velocity profile

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3. RESULTS

a. Heat gain across heat exchanger

$$1\frac{kj}{sec} = 860.421\frac{Kcal}{hr}$$
$$Q = 985.106\frac{kj}{sec}$$
$$Q = 847605.8896\frac{Kcal}{hr}$$

b. Total saving in rupees per year (I-coal)

Heat gain across heat exchanger Total saving of coal = GCV of coal 847605.8896 Total saving of coal = 3400 Total saving of coal = 249.2958 kg/hrTotal saving of coal = 0.2492958 ton/hr Total saving of rupees per hour = Coal Rupees per ton \times Tatal saving of coal ton per hour $= 3500 \times 0.2492958$ Rs per hr = 872Total saving of rupees per day = Tatal saving of rupees per hour $\times 24$ $= 872 \times 24$ Rs per day = 20928Total saving of rupees per Month = Tatal saving of rupees per day \times 30 $= 20928 \times 30$ Rs per month = 627840Total saving of rupees per year = Tatal saving of rupees per month \times 12 $= 627840 \times 12$ Rs per year = 7534080

4. CONCLUSION

a. Heat transfer rate of flue gas of installed new heat exchanger (Q_2)

 $Q = m_{fg} \times c_{p (fg)} \times (T_{h_1} - T_{h_2})$ $Q = 44.90 \times 1.097 \times (473 - 453)$ Q = 985.106 kj/sec

b. Total heat transfer rate of flue gas of WHRB (Q₁)

 $\begin{array}{l} Q = m_{fg} \times c_{p~(fg)} \times (T_{h_1} - T_{h_2}) \\ Q = 44.90 \times 1.332 \times (1223 - 373) \\ Q = 50454.13 \ \text{kj/sec} \end{array}$ Heat transfer rate of flue gas (Q) = $\displaystyle \frac{Q_1 - Q_2}{Q_1} \times 100$ Heat transfer rate of flue gas s (Q) = $\displaystyle \frac{50454.13 - 985.10}{50454.13} \times 100$ Heat transfer rate of flue gas (Q) = 98 % Utilised of heat transfer rate of flue gas = 100 - Heat transfer rate of flue gas (Q) Utilised of heat transfer rate of flue gas = 2%

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Payback period on the basis of I coal c.

Payback period = $\frac{\text{Total project cost}}{\text{Actual profit per day}}$ Payback period = $\frac{185000}{20928}$ Payback period = $\frac{185000}{20928}$ Payback period = 8.83 Days Payback period = 9 Days

d. Water tube heat exchanger successfully designed for WHRB

- Waste heat is used from WHRB.
- As suggested by energy auditor that, 5% total heat is remains unutilized between ESP and • economizer, it can be utilize by placing heat exchanger in between.
- We analyzed that 5% of flue gases is wasted from the WHRB. So, we suggested for installation of . a water tube heat exchanger to recover a waste heat up to 2%.
- Water tube heat exchanger (Inline arrangement and cross flow type).
- It is a vertical and bends in U shape. •

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